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Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

Theoretical Investigation of Interaction of Hydrogen Dithiophosphate with 2-Phenyl-2H-1,2,3-diazaphosphole

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Online publication date: 27 October 2010

To cite this Article Khusainova, Narkis , Zverev, Vladislav , Garipova, Goulnara , Cherkasov, Rafael and Pudovik, Arkady(2002) 'Theoretical Investigation of Interaction of Hydrogen Dithiophosphate with 2-Phenyl-2H-1,2,3-diazaphosphole', Phosphorus, Sulfur, and Silicon and the Related Elements, 177: 8, 2229

To link to this Article: DOI: 10.1080/10426500213307 URL: http://dx.doi.org/10.1080/10426500213307

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Phosphorus, Sulfur and Silicon, 2002, Vol. 177:2229 Copyright © 2002 Taylor & Francis 1042-6507/02 \$12.00 + .00

DOI: 10.1080/10426500290095269



THEORETICAL INVESTIGATION OF INTERACTION OF HYDROGEN DITHIOPHOSPHATE WITH 2-PHENYL-2H-1,2,3-DIAZAPHOSPHOLE

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(Received July 29, 2001; accepted December 25, 2001)

We showed previously that the cyclic two-coordinate phosphorus derivative—5-methyl-2-acetyl(phenyl)-2H-1,2,3-diazaphosphole takes up 0,0'-dialkyl hydrogen phosphorothio- and dithioates across the $\sigma^2 \lambda^3 P = C$ bond; the anionic moiety of the thio and dithio acid was directed toward the two-coordinate phosphorus atom. ¹ In the ³¹P NMR spectra of the resulting 3-(dialkoxyphosphinoyldithio)-2-phenyl-1,2,3-diazaphospholines 1 the chemical shifts of the three and four coordinate phosphorus atoms were equivalent owing to intramolecular interaction of the P(III) atom with the thione sulfur atom in the adduct. As a continuation of these studies, the electron and molecular structure of adduct of hydrogen dimethyl dithiophosphate with 2phenyl-5- methyl-2H-1,2,3-diazaphosphole, containing P=C bond, have been studied by the methods of quantum chemical calculations (MNDO, PM3, ab initio) with the usage of the photoelectron spectrosopy. It has been shown, that the equality of the shifts of P(III) and P (IV) in 31P NMR spectra of the adduct 1 probably takes place in the results of the real of the structure with nearly equal lengths of four P—S bonds.

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This work was supported by grant of the Russian Foundation for financial Support of the Leading Scientific School, by the program "The Universities of Russia—Basic Researches" and by the joint program of CRDF and Russian Ministry of Education (REC-007).

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